



# Tar analysis methods for small scale gasification systems

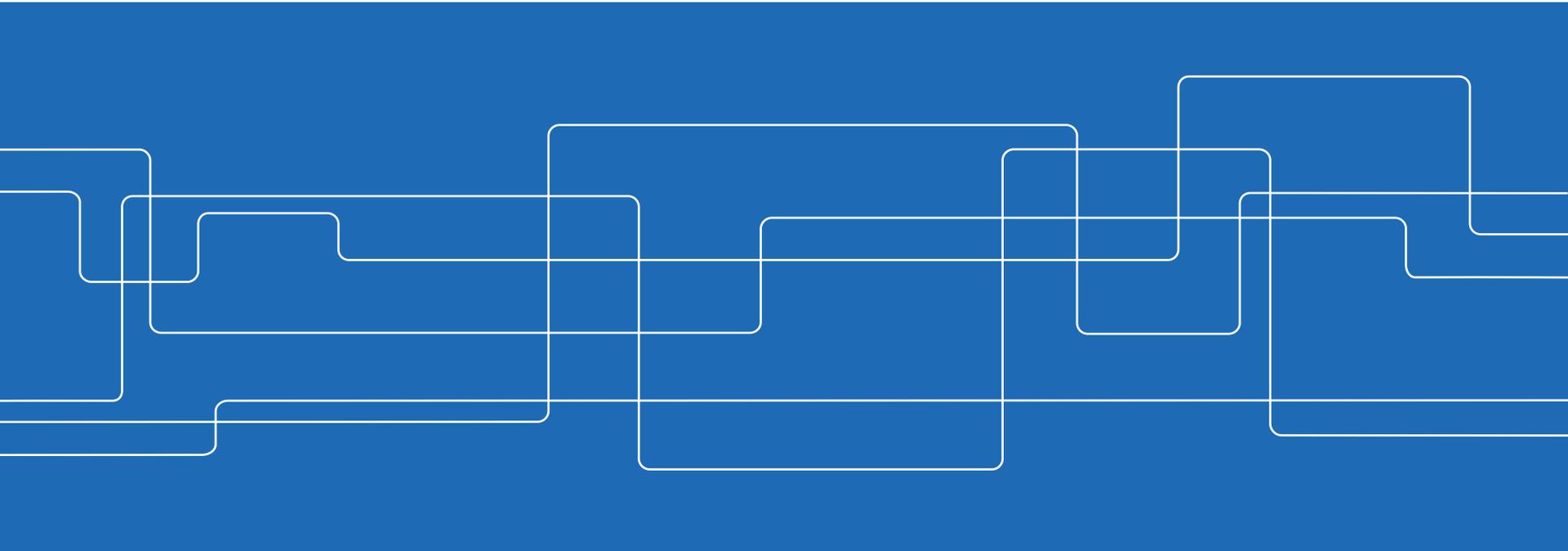
*...from a KTH perspective...*

**Klas Engvall**

KTH Royal Institute of Technology

Dept. of Chemical Engineering and Technology

Stockholm, Sweden





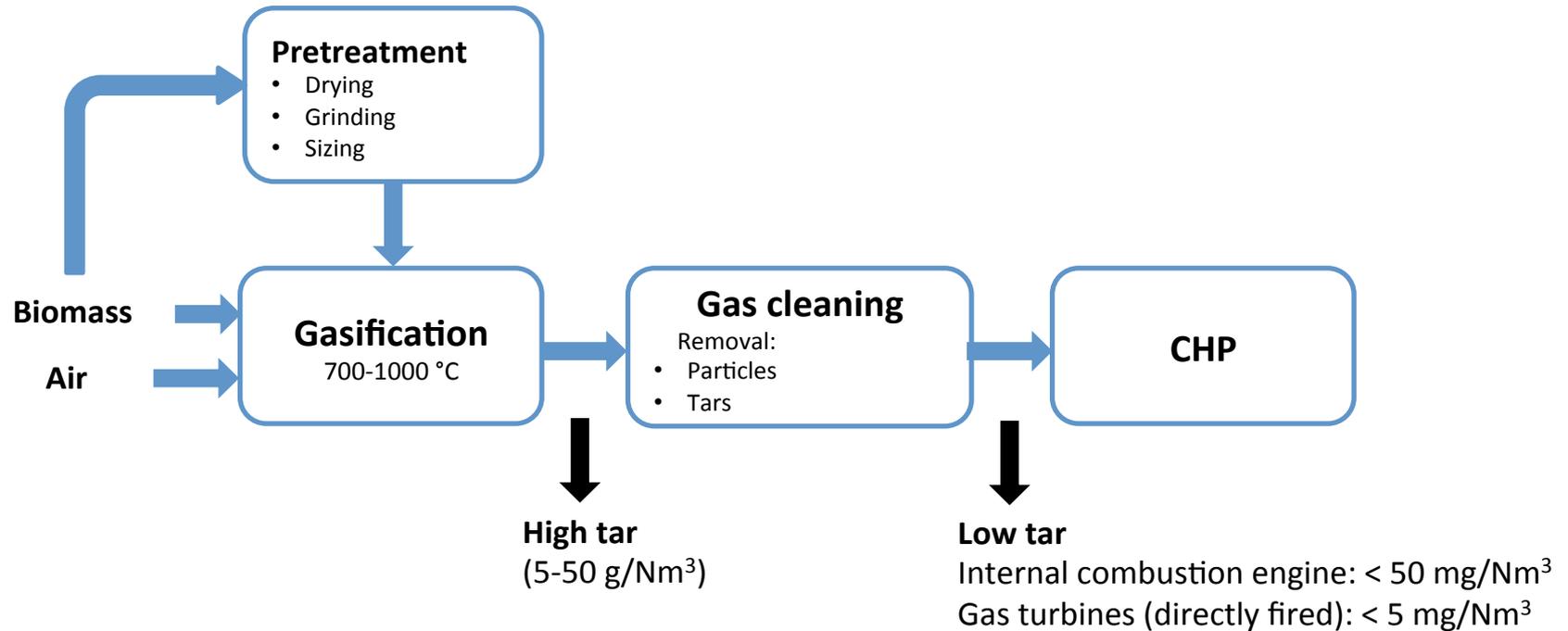
# Outline

- Background
- Tar analysis in a small scale gasification system
  - Offline methods
    - “Conventional” tar analysis
    - Solid Phase Absorption (SPA)
    - SPME
  - Online methods
    - FID
    - PID
  - What is the conclusion?
- Summary



# Background

# Small scale gasification systems



# Challenge biomass gasification tars

- Production of condensable polyaromatic “tars” is inherent in most biomass gasification processes
- Tars foul and can plug equipment downstream of the gasifier
- Challenging to remove from the produced gas
- Reduce energy efficiency of gasification process
- Reports of as much as 10 % of biomass carbon ending up in the form of tars



# Challenge biomass gasification tars

Example internal combustion engine



Throttle valve tar deposits with water scrubbing (left) ,  
oil scrubbing (right)



# What is tar?

The term "tar" is vague and the definition vary.

One definition is "organic molecules with a molecular weight higher than that of benzene" ( $M_w = 78 \text{ g/mol}$ ).



# What is tar?

## Light tars

Organic compounds that can be analysed with GC as well as HPLC. (Mw 79-300 g/mol). They are **volatile and semi-volatile aromatics and phenolics**.

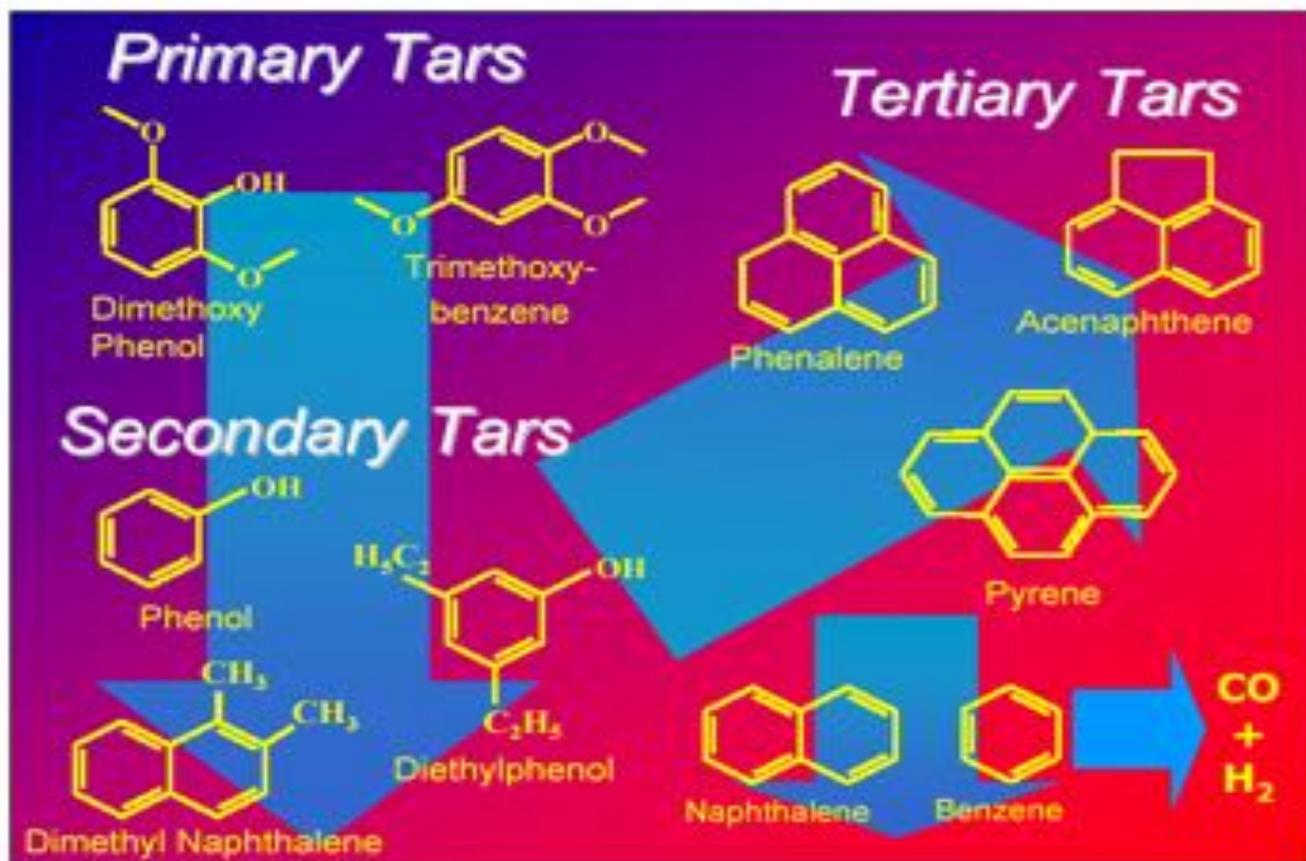
## Heavy tar

Organic compounds with so high boiling points that they can be analysed only by HPLC, not with GC. **They are mixtures of high molecular weight "non-volatile" polar compounds (Mw  $\approx$  >300 g/mol)**

Total tar = sum of light and heavy tar

# What is tar?

Polycyclic aromatic compounds



Oxygenated compounds

Phenolic compounds and olefins



# General scheme tar analysis

A few common steps:

1. **Sampling of the tar:** Generally collected from a side stream, including more or less complicated sampling equipment to attain a representative sample.
2. **Storage of sample:** Only valid for offline methods.
3. **Pre-preparation/conditioning of the sample:**  
Offline methods: The collected tars are extracted to or dissolved in an appropriate solvent for further chemical analysis.  
Online measurements: Conditioning such as drying of gas removal of particulates etc. may be required depending on the analytical technique.
4. **Analysis of sample:** Chemical analysis of pre-prepared/conditioned tar sample. Most common analytical techniques are based on gas chromatography (GC) or high pressure liquid chromatography (HPLC).

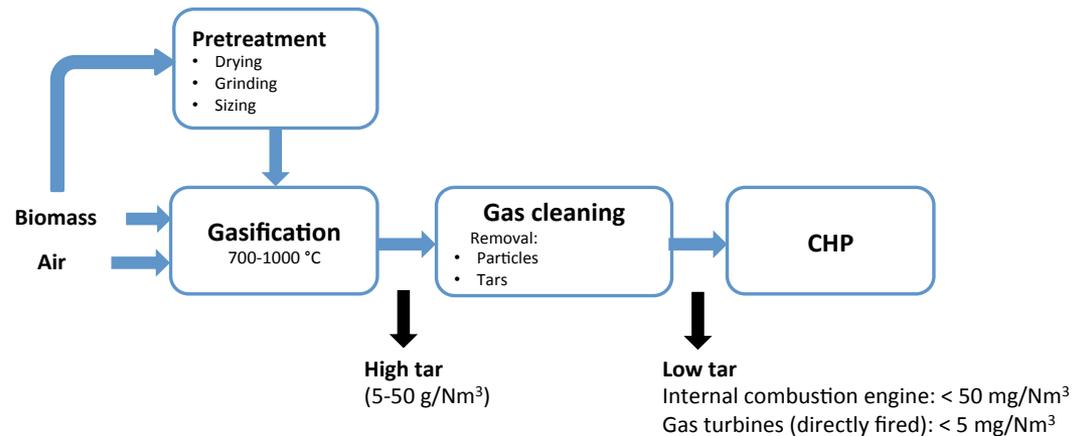
# All these methods!

Method	Development status	Application	Measured data	Offline/ Online	Sampling and analysis time	Cost
Tar protocol/ guideline	CEN/TS pre-standard	Laboratory use	Gravimetric tar (Class 1) GC-FID (Class 2-5)	Offline	Long sampling and analysis time	Very expensive
Petersen column	Portable device	Laboratory use	Gravimetric tar (Class 1-5)	Offline	Long sampling and analysis time	Low
GC-FID and GC-MS	Not easy to transport	Laboratory use	Individual tar compounds (Class 2-5)	Offline/ Online	Short sampling time and relatively short analysis time	Expensive
HPLC	Not easy to transport	Laboratory use	Individual tar compounds (Class 1-5)	Offline/ Online	Long sampling and analysis time	Expensive
SPA	Easy to use probe	Laboratory use	Individual tar compounds (Class 2-5)	Offline	Relatively short sampling and long analysis time	Expensive
SPME	Easy to use probe. Under development	Laboratory use	Individual tar compounds (Class 2-5)	Offline	Relatively short sampling and long analysis time	Low
Online tar analyser]	Portable device	Industrial use	Individual tar compounds (Class 2-5)	Online	Relatively short sampling and analysis time	Expensive
MBMS	Transportable	Industrial use	Individual tar compounds (Class 2-5)	Online	Short sampling and analysis time	Very expensive
GC/LAMS	Transportable	Laboratory use	Individual tar compounds (Class 2-5)	Online	Short sampling time and relatively short analysis time	Very expensive
PID	Under development	Industrial use	Individual tar compounds (Class 2-5)	Online	Short sampling and analysis time	Low
Raman spectroscopy	Not transportable	Industrial use	Gravimetric tar (Class 1-5)	Online	Short sampling and analysis time	Very expensive

- Large number of methods for tar measurement
- Generally:
  - For research and laboratory use
  - Complex – needs expertise
  - Several are expensive
  - Not robust enough process
  - ...

# What method to use?

- Depends on information desired!
  - Qualitative or quantitative information?
  - Information about chemical composition?
  - R&D or industrial monitoring?
- Reliability?
- Costs?
- ...



# Requirements tar analysis small scale systems

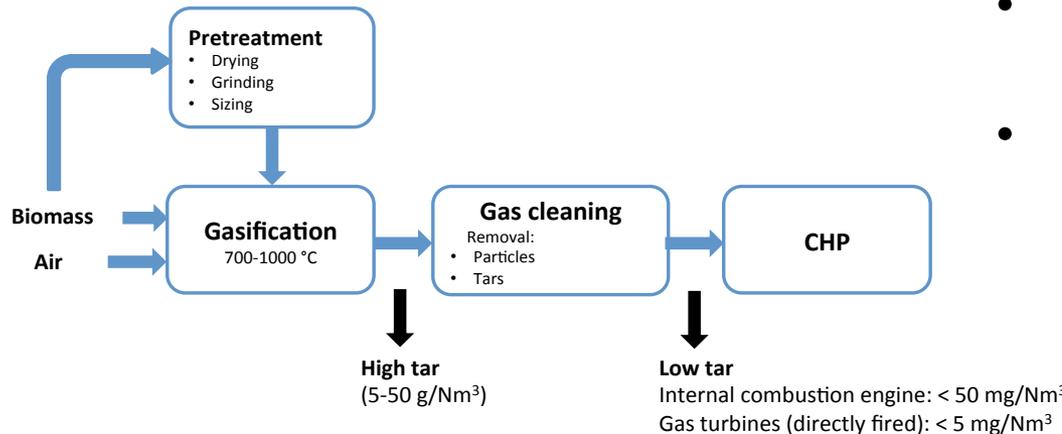
## Research

No real requirements

“A matter of need to know and costs!”

## Industrial monitoring/analysis

- Reliable - high repeatability
- Low staffing
- No or low need for expertise
- Low costs
- Fast
- Tar composition generally not of interest
- Preferably online





# Tar analysis in a small scale gasification system



# Offline methods

“Conventional” tar analysis

**Tar protocol** developed over several projects supported by IEA Bioenergy Task 33, US DOE and European Commission 1998-2005

- Significant contributions by ECN, VTT, KTH, DTI, BTG, NREL

Adopted as CEN standard for tar sampling



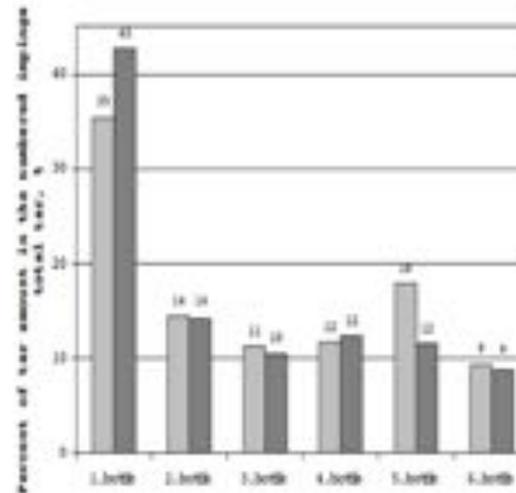
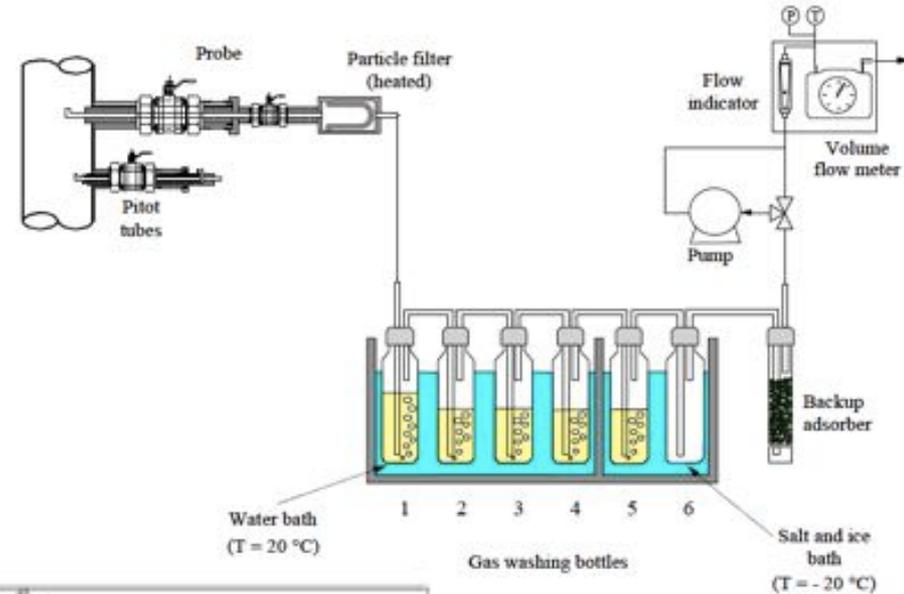
# Offline methods

## “Conventional” tar analysis

### Procedure

- Draw specific volume of process gas through a filter and then a series of cold impingers to collect the tars
- Evaporate solvent to measure **gravimetric** (total) tars
- Analyze tars by **GC-MS** to **evaluate composition**

Quantitative, but very laborious





# Offline methods

## Solid Phase Adsorption (SPA)

- SPA sampling and analysis method was developed by KTH in the 1990's.
- SPA used for measurement of the concentration (mass) of individual light aromatic hydrocarbons and phenols.
- The SPA-method is restricted to GC-available (GA) compounds only.
- These compounds are, however, significant process markers that provide good measures of reactor performance and gas quality.
- At  $T = 900^{\circ}\text{C}$  and above the GA-compounds roughly correspond to the total tar amount.

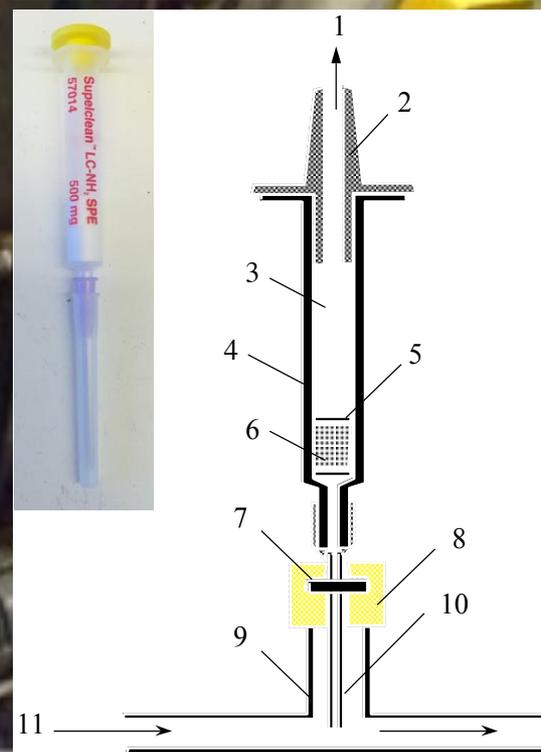


# Offline methods

## Solid Phase Adsorption (SPA)



Solid Phase Extraction (SPE)  
NH<sub>2</sub> tube



1 = to syringe or electrical pump; 2 = adapter (polypropylene); 3 = sample reservoir; 4 = sorbent tube (PP, 1.3 OD x 7.5 cm); 5 = fritted disc (20 mm polyethylene); 6 = amino-phase sorbent (40 mm, 60 Å); 7 = rubber/silicone septum; 8 = septum retainer (polypropylene); 9 = "Tee"-adapter (glass); 10 = syringe needle (stainless steel); 11 = producer gas.

# Offline methods

## Solid Phase Adsorption (SPA)



### Sampling

“T”, needle, SPE-NH<sub>2</sub> tube and 100 ml syringe.

Custom made reversible SPE tube.

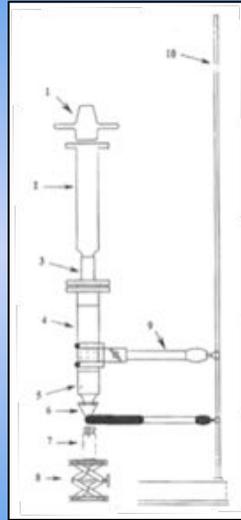
Sampling of 100 ml in 1 min.



### Sample storage

The SPE tube is capped in both ends after sampling.

Samples stored in a fridge/freezer



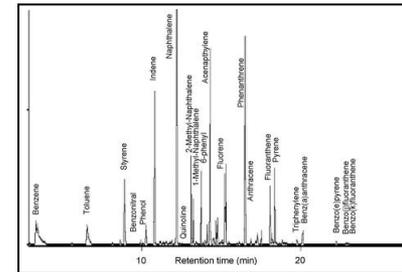
### Sample preparation

Elution for aromatic and phenolic compounds



### Chemical analysis

GC-FID - Gas chromatograph with flame ionisation detection



### Results

Detection limit: 2.5 mg/Nm<sup>3</sup> (for detectable tars)



# Offline methods

## Challenges of SPA Method

- Inleakage of air, especially for sub-atmospheric pressure systems
- Using temperature high enough to avoid tar condensation yet low enough not to melt septum
- Plugging of needle by septum material
- Condensation of tars in needle of syringe
- Undesirable heating of SPE column during sampling due to temperature, steam condensation
- **Breakthrough of light tars (BTX)**
- **Desorption of light components from SPE cartridge during storage**
- Efficient elution of aromatic and phenolic compounds
- **Inability to measure heaviest tars**
- **Consistency of procedures for sampling and analysis**



# Offline methods

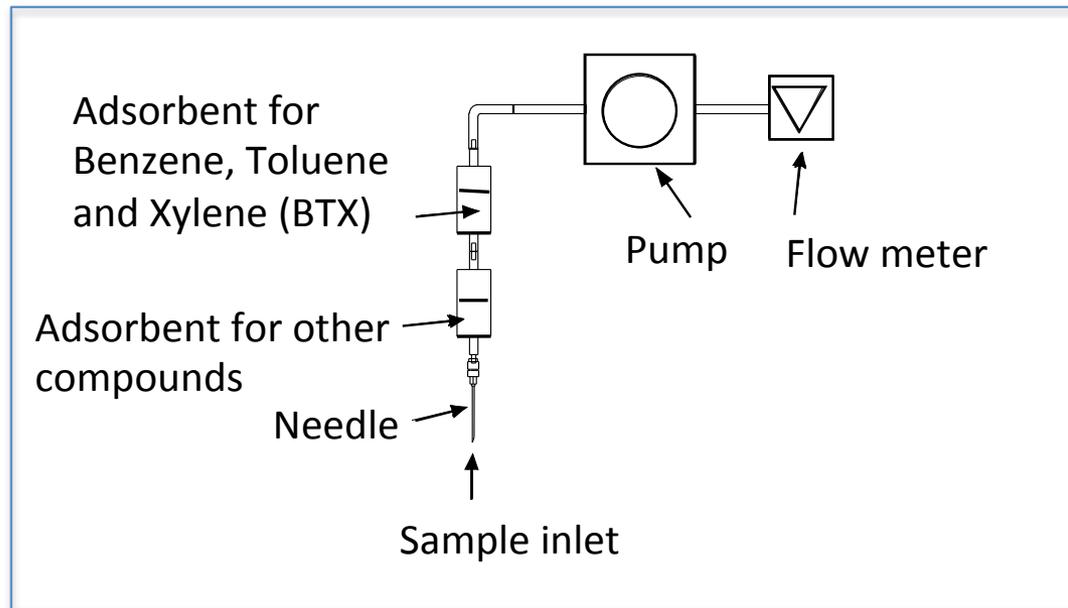
A (biased?) comparison between SPA and “Tar guideline”

<b>Cold solvent trapping (CST)</b> <b>(“Tar guideline”)</b>	<b>Traditional SPA</b> <b>(“KTH”)</b>
<p><b>Advantages:</b></p> <ul style="list-style-type: none"><li>- Gives total tar, heavy and light tar</li></ul> <p><b>Drawbacks:</b></p> <ul style="list-style-type: none"><li>- Time consuming, sampling as well as analysis</li><li>- Large solvent volumes</li><li>- Not suitable for (very) low tar concentrations</li><li>- Low precision</li></ul>	<p><b>Advantages:</b></p> <ul style="list-style-type: none"><li>- Uncomplicated and fast sampling</li><li>- Low cost</li><li>- High accuracy and reproducibility</li><li>- Sampling and analysis can be done separately</li></ul> <p><b>Drawbacks:</b></p> <ul style="list-style-type: none"><li>- Not suitable for heavy tars</li><li>- B(TX) must be analysed within a few hours</li></ul>

# Offline methods

SPA analysis at low tar concentrations and separate BTX analysis (KTH)

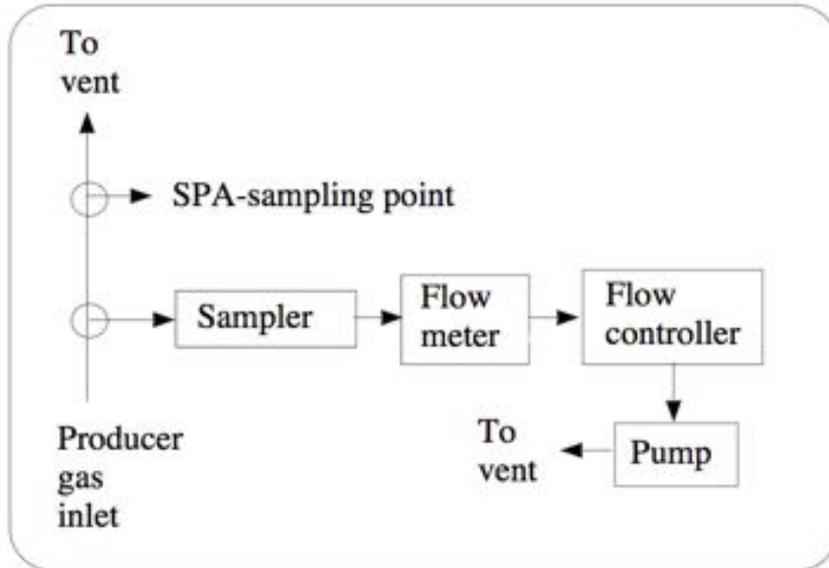
Determination of light tar in low concentrations



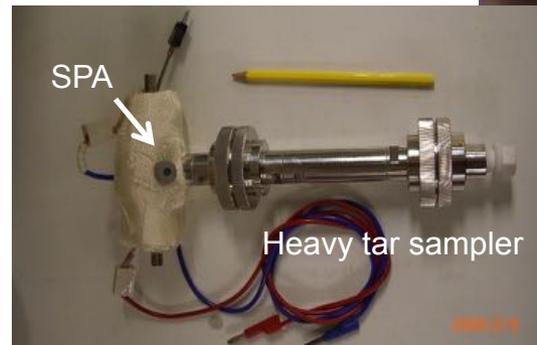
*“A few other variants exists, e.g. Chalmers and ECN”*

# Offline methods

SPA analysis combined with gravimetric tar (KTH)



Heated and isolated “T”-connection with SPA-septa (left) and heavy tar sampler (right)

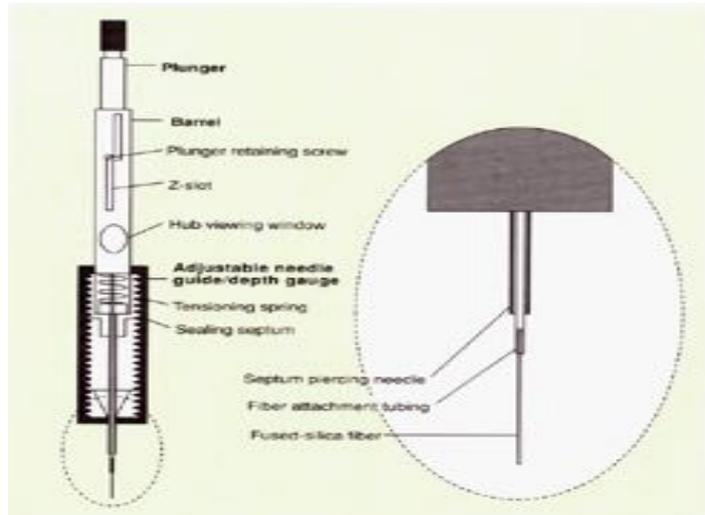


Sampler mounted on atmospheric fluidised bed gasifier

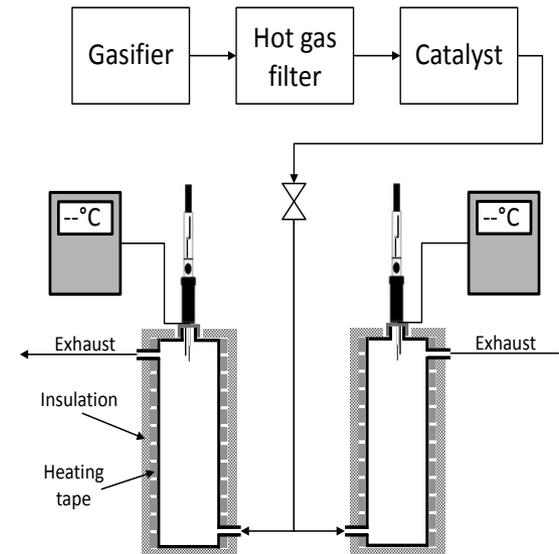
# Offline methods

## Solid Phase Micro Extraction (SPME)

### SPME device



### Experimental set-up KTH

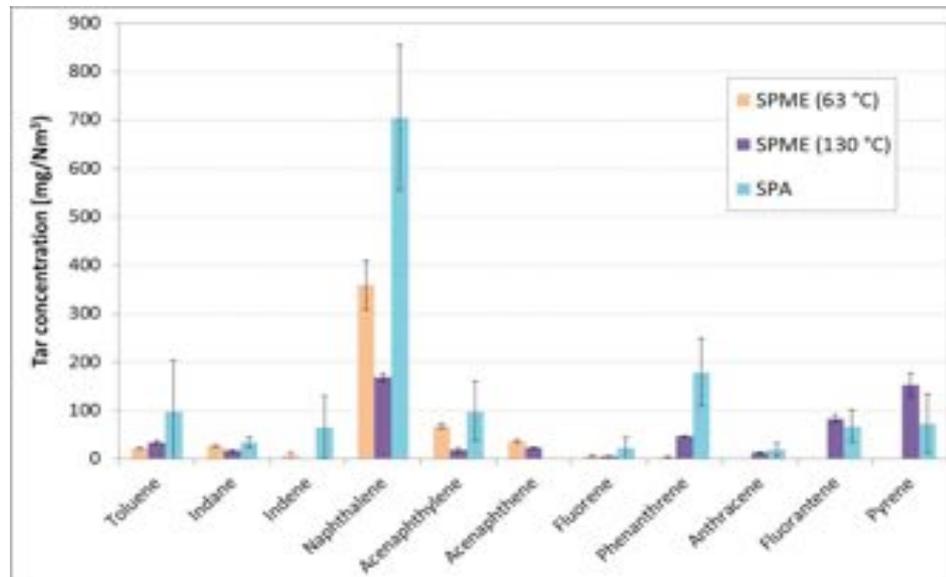


- Method under development
- Extraction of analytes from a sample matrix onto a stationary phase (non polar) - silica fibre with 50  $\mu\text{m}$  polydimethylsiloxane (PDMS)
- Desorption of the analytes in an analytical instrument (GC).
- Developed for low tar content analysis in e.g. syngas applications

# Offline methods

## Solid Phase Micro Extraction (SPME)

- Tests on real gas with sampling time 10 min.
- The results showed that SPME method is a fast and accurate for low tar concentrations
- Analysis at trace levels below  $0.1 \text{ mg/Nm}^3$  (e.g., syngas production) will be possible at  $60 \text{ }^\circ\text{C}$  for all compounds heavier than naphthalene



# Online methods

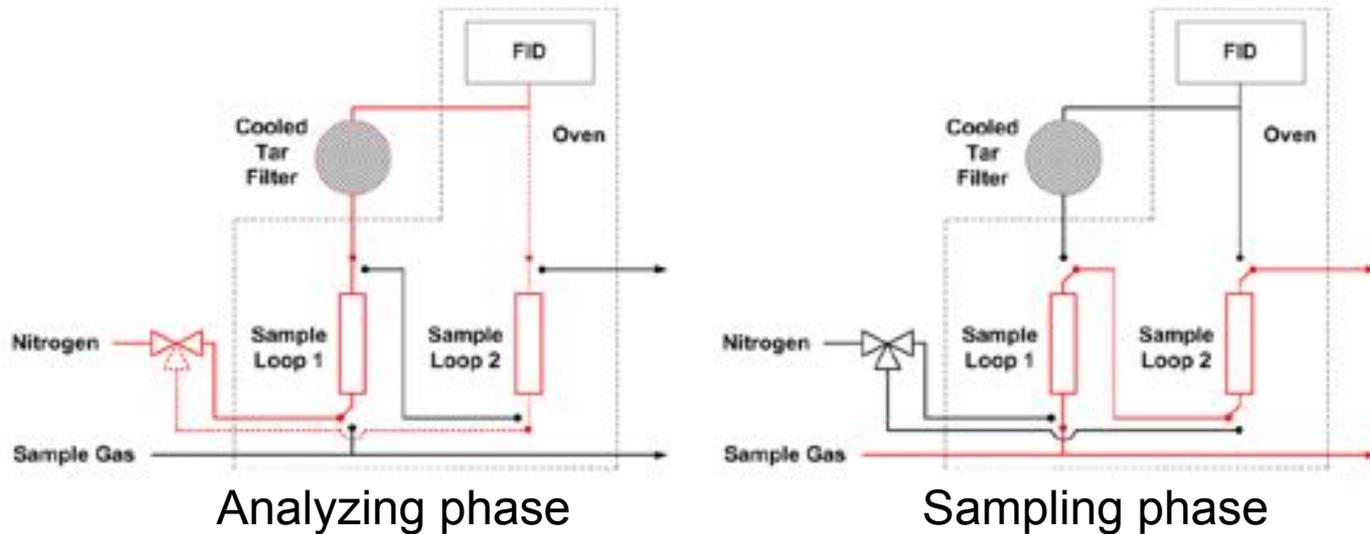
## Flame Ionization Detector (FID)



- Prototype developed by researcher at University of Stuttgart.
- The instrument is using FID as a the detector principle.
- The instrument can determine:
  - Total hydrocarbon concentration
  - The non-condensable hydrocarbon concentration
  - The tar concentration

# Online methods

## Flame Ionization Detector (FID)



### Measurement principle:

Difference measurement of the organically bound carbon in the sample gas of two sample loops with equal volume.

# Online methods

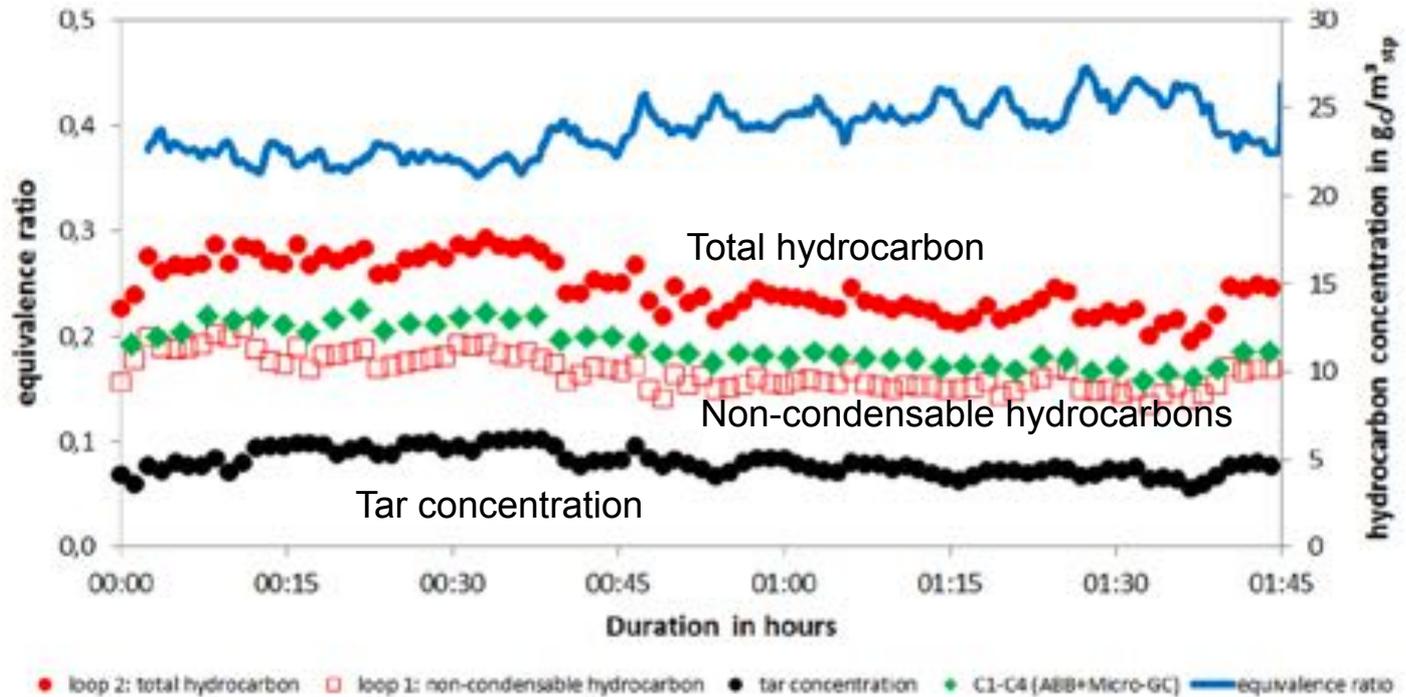


Fig. 9 Results of first field test with real producer gas from gasification

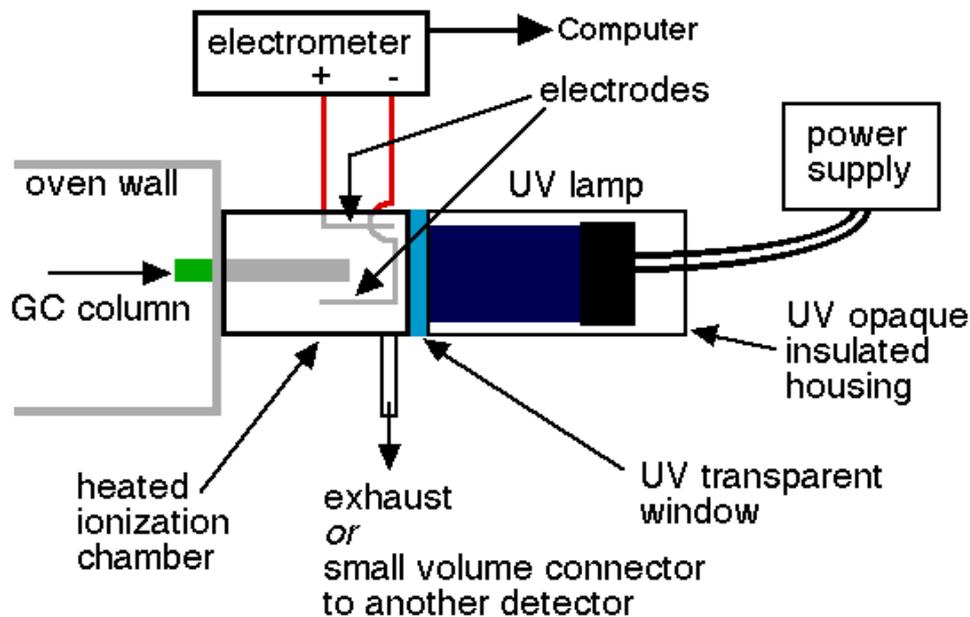
## General impression:

- Easy to use
- Provide accurate results in comparison with "Tar protocol"
- The choice of a suitable tar filter material for the difference measurement was identified as one of the major challenges to gain realistic results.

# Online methods

## Photo Ionization Detector (PID)

*"Method under development at KTH"*



A molecule with an ionization potential (IP) lower than the actual energy ( $E = h\nu$ ) of a photon is ionized.



Energy required to remove an electron is different for each compound.

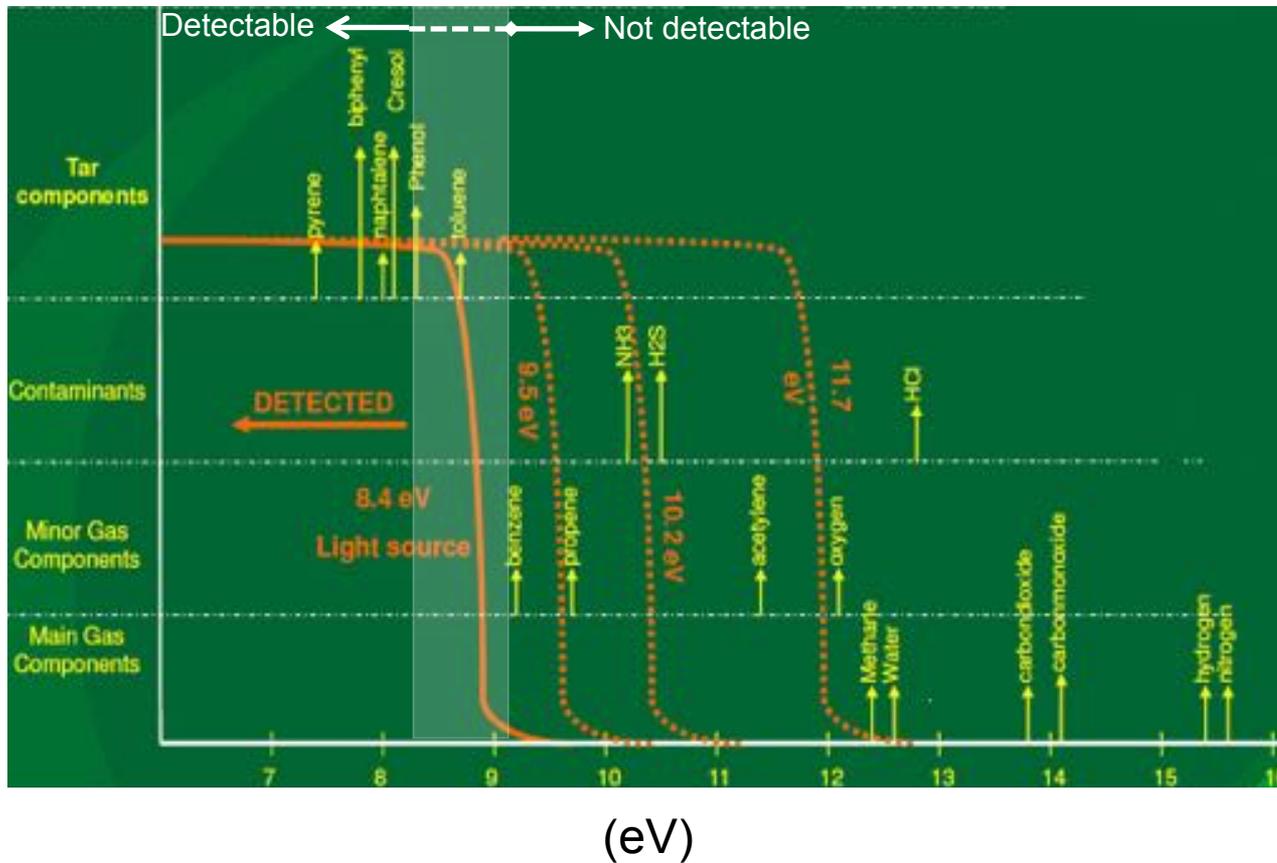
Typical tar compounds require relatively little energy



Possibly a selectivity can be achieved.

# Online methods

## Photo Ionization Detector (PID)



Energy of the light depends on the gas inside the lamp

Xenon = 8.4 eV

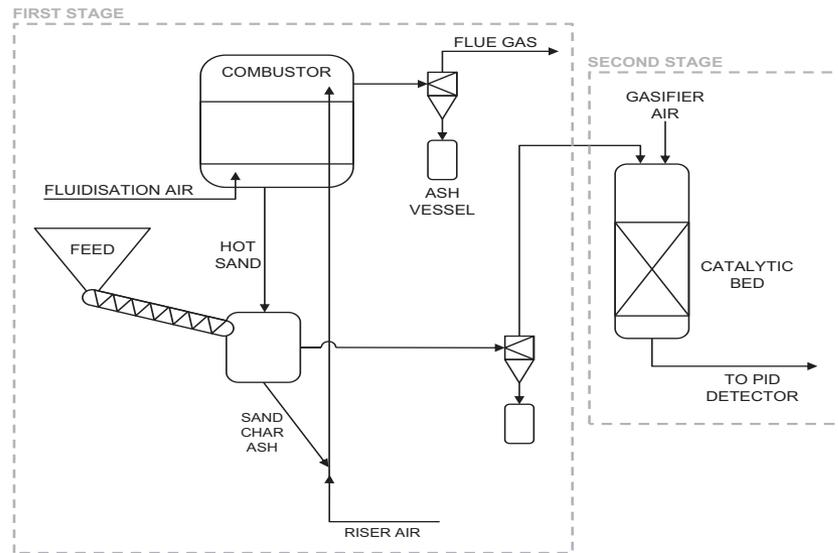
Aromatic compounds with IP < 8.4 eV can be detected, e.g.:

- Naphthalene
- Acenaphthene
- Fluorene
- Anthracene
- Pyrene

# Online methods

## Photo Ionization Detector (PID)

### PID in a real gas stream

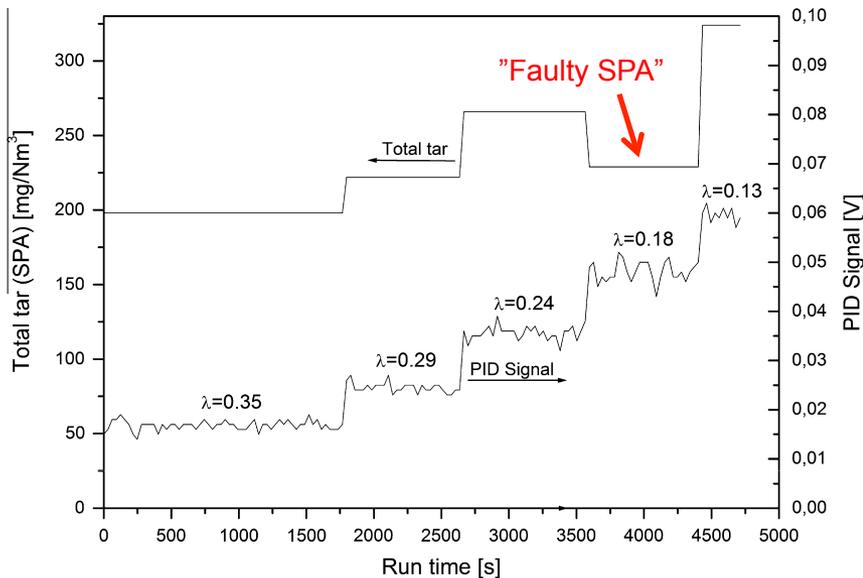


Schematics of the tar measurement system for real gas tests at BTG

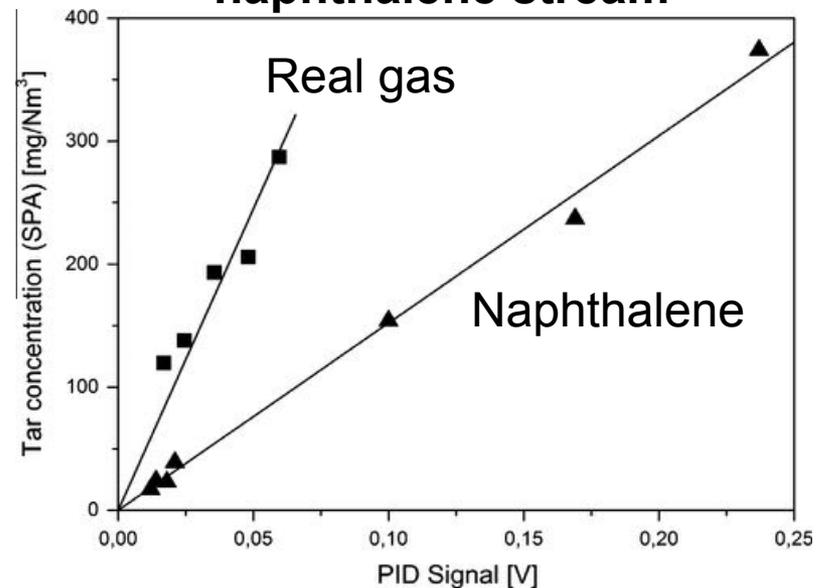
# Online methods

## Photo Ionization Detector (PID)

### PID in a real gas stream



### Real gas stream vs pure naphthalene stream



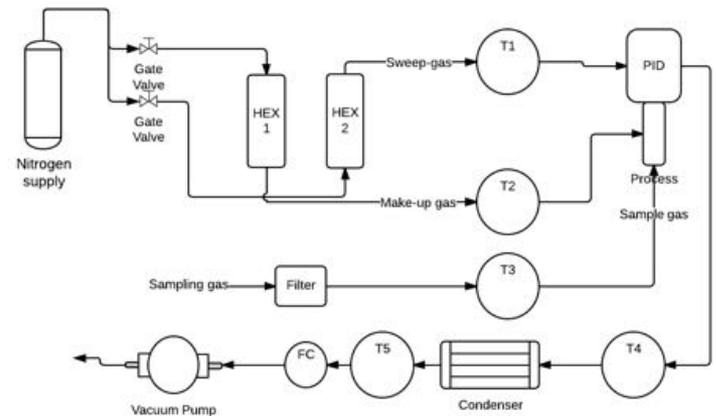
- The PID signal follows the SPA signal at almost all different tar levels
- Both real gas and naphthalene PID signal shows a linear correlation comparing with SPA tar content

# Online methods

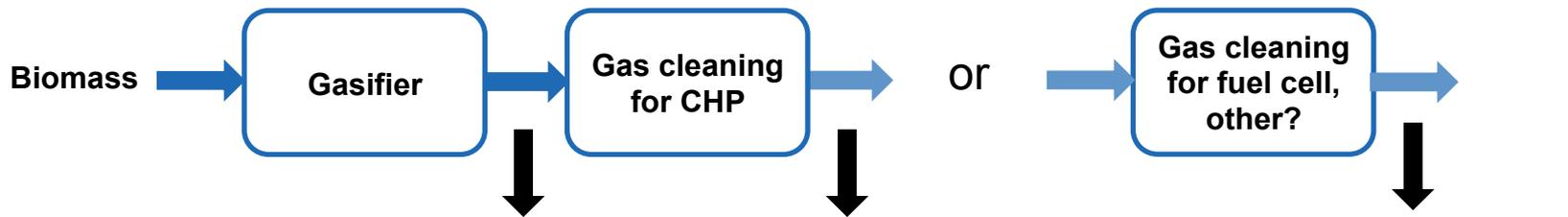
## Photo Ionization Detector (PID)

### Some observations:

- Different response curves of the compounds will make quantification less accurate during analysis of real producer gas
- Fouling of UV lamp window with time is an issue.
- PID prototype test system developed to address the window fouling problem
- Prototype tests in progress



# What is the conclusion?



	High tar (10-50 g/Nm <sup>3</sup> )	Medium tar level (5-50 mg/Nm <sup>3</sup> )	Low tar Below dew point: 0.1 mg/Nm <sup>3</sup> at 100 °C
<b>Offline</b>	<ul style="list-style-type: none"> <li>• SPA</li> <li>• Tar protocol</li> </ul>	<ul style="list-style-type: none"> <li>• SPA</li> <li>• Tar protocol</li> </ul>	<ul style="list-style-type: none"> <li>• SPME</li> </ul>
<b>Online</b>	<ul style="list-style-type: none"> <li>• PID</li> <li>• FID</li> </ul>	<ul style="list-style-type: none"> <li>• PID</li> <li>• FID</li> </ul>	No viable technique



# Summary



# Summary

- Simple, low-cost, yet robust means of measuring and characterizing tars is desirable, especially for small scale systems
- Impinger-based method of standard tar protocol is relatively robust but time consuming and laborious
- SPA method much simpler and equally as good for many situations, but does have drawbacks
- Continued development of SPA procedure will improve robustness and utility of the method
- Developed FID and PID under development are both promising candidates for future industrial online tar monitoring



# Thank you!

E-mail: [kengvall@kth.se](mailto:kengvall@kth.se)